comparison. It appears that the uniform central part of the integrated spot is quite long, indicating that the original reflexions had a flat-topped distribution and that the total translation could have been reduced. Table 1

Table 1

	Non-integrated		Integrated	
θ (°)	Top	Bottom	Top	Bottom
16	56	22	20	20
20	40	20	14	12
30	6	4	2	3
32	28	16	9	10
34	22	12	7	7
40	28	22	10	16
42	11	7	4	3
44	18	11	5	5
50	11	8	4	5
70	20	14	8	7
71	8	6	4	3

gives the visually estimated intensities of the reflexions on both halves of the two films. (The c axis of calcium dithionate tetrahydrate is an axis of symmetry.)

Although the method can be used directly by adjusting the camera between exposures it would not be difficult to devise a simple mechanism which could be fitted to existing cameras to give the camera the required incremental translation at the end of each traverse.

The proposal of Whittaker (1953) to uncouple the oscillation and translation motions of the Weissenberg goniometer, and allow the camera to oscillate (through a small distance) parallel to the axis independent of the crystal oscillation, would achieve the same result in providing a linear integration of the spots. The integrated spots would be short streaks perpendicular to the single layer line and some of the advantages of the Weissenberg photograph would be lost.

I am grateful to Dr D. C. Phillips for giving me access to the manuscript of his theoretical work on this subject, prior to publication, which led to this method.

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Re-investigation of the crystal structure of CsI₃. By H. A. Tasman and K. H. Boswijk, Laboratorium voor Anorganische en Physische Chemie der Rijks Universiteit, Bloemsingel 10, Groningen, The Netherlands

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A recent X-ray examination of Cs₂I₈ showed this compound to contain I₃ ions in weak interaction with I₂ molecules (Havinga, Boswijk & Wiebenga, 1954). It was thought interesting to compare the dimensions of the I_3^- ion constituent in Cs_2I_8 with those of the I_3^- ion in tri-iodides. It appeared, however, that no data on the tri-iodides were available which could be expected to be sufficiently accurate for this purpose. For CsI₃ only unit-cell dimensions and space group have been determined (Bozorth & Pauling, 1925). The structure of NH₄I₃, isomorphous with CsI₃, was investigated by Mooney (1935). The I_3^- ion was reported to be linear to within 3°; the two I-I distances in the ion were found to be different (2.82 and 3.10 Å). According to the author herself, however, this difference might be fictitious, the analysis being based only on qualitatively estimated intensities, making use of trial-and-error methods. It was therefore decided to re-investigate the crystal structure of one of the tri-iodides. CsI3 was chosen because, with this compound, an accurate determination of both anion and cation coordinates should be possible.

Making use of NaCl powder diagrams superimposed on rotation photographs, the lattice constants were determined:

$$a = 6.86$$
, $b = 9.98$, $c = 11.09$ Å ($\lambda(Cu K\alpha) = 1.5418$ Å).

Values for these unit-cell dimensions, derived independently from various photographs made in different cameras, showed a standard deviation of approximately 0.01 Å. The lattice constants are in fair agreement with those reported by Bozorth & Pauling (1925).

The characteristic absences (h0l for l odd and hk0 for

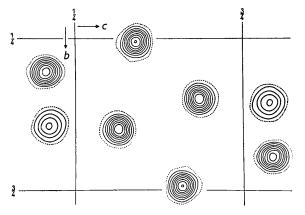


Fig. 1. Electron-density projection along [100]. Contour intervals: 10 e.Å⁻² (with 10-electron line dotted) except on cesium ions (heavy lines), where the interval is 20 e.Å⁻².

h+k odd) indicate the space group Pmcn. Since, even on photographs of very long exposure, no exceptions to these absences were observed, and since the unit cell contains only 4 molecules, all atoms must have x coordinates $x=\frac{1}{4}$ and $\frac{3}{4}$.

This means that the remaining coordinates can be obtained from the [001] projection only.

Ordinary and integrated Weissenberg photographs were made about the a axis of a roughly isodimensional crystal, approximately 0.0075 cm. in diameter, using both Mo and Cu radiation. The intensities of 98 independent 0kl reflexions were obtained and corrected for absorption and Lorentz- and polarization factors. Starting from the coordinates as given by Mooney for NH₄I₃, the [100] projection was refined by a series of consecutive Fourier syntheses. The final contour map is shown in Fig. 1. The coordinates were corrected for the finite-series effect by the 'back-shift' method. They are listed below.

	$oldsymbol{y}$	z	
Cs	0.4640	0.8259	
I_1	0.3598	0.1641	
I_2	0.5500	0.3778	
I_2 I_3	0.7367	0.5697	

With these coordinates the value of $R=\mathcal{L}\{|F_o|-|F_c|\}$; $\mathcal{L}[F_o]$ is 0·105, the summations being taken over all observed reflexions. In the calculated structure factors an isotropic temperature factor $\exp{[-1\cdot00~(\sin{\theta/\lambda})^2]}$ has been introduced.*

The dimensions of the I_3^- ion may be read from Table 1; those for $\mathrm{NH_4I_3}$ and $\mathrm{Cs_2I_8}$ have been included for comparison. The bond lengths and angles observed in the early analysis of $\mathrm{NH_4I_3}$ are in surprisingly good agreement with our results.

Table 1. Dimensions of I_3^- ion

	NH_4I_3	$\mathbf{CsI_3}$	$\mathrm{Cs_2I_8}\dagger$
I_1-I_2 (Å)	2.82	$2.83_0(0.01_5)$	$2.82_{5}(0.01_{8})$
I_2-I_3 (Å)	3.10	$3.03_{5}(0.01_{5})$	$3.00(0.01_8)$
$I_1 - I_2 - I_3$ (°)	177	176.3(0.5)	176.8(0.8)

Values in parentheses represent the standard deviations calculated with Cruickshank's formula (Cruickshank, 1949); an estimated allowance for imperfect correction for finite series was included.

* A table of observed and calculated structure factors has been deposited as Document No. 4431, with the ADI Auxiliary Publications Project, Photoduplication Service, Library of Congress, Washington 25, D.C., U.S.A. A copy may be secured by citing the Document number and by remitting \$1.25 for photoprints, or \$1.25 for 35 mm. microfilm. Advance payment is required. Make cheques or money orders payable to: Chief, Photoduplication Service, Library of Congress.

† The values listed for $\operatorname{Cs}_2 I_8$ are slightly different from those given in the article on the structure of this compound (Havinga, Boswijk & Wiebenga, 1954) because a more accurate correction for the finite-series effect was made after the publication of this paper.

From the data given in Table 1 it is beyond doubt that in CsI_3 and Cs_2I_8 the I_3^- ion has an asymmetric, nonlinear structure. It is difficult to judge, however, whether the rather small deviation from linearity is a property of the 'free' ion itself or may be explained by the influence of intermolecular forces in the crystals.

Fig. 2 gives a survey of the interatomic distances. The Cs-I distances indicated amount to 3.92 Å on an average; they are comparable with those observed in CsI (3.95 Å) and Cs₂I₈ (3.85-4.07 Å). The short intermolecular I–I distances (4.34 Å on an average) may be compared with those in NH₄I₃ (4.28 Å).

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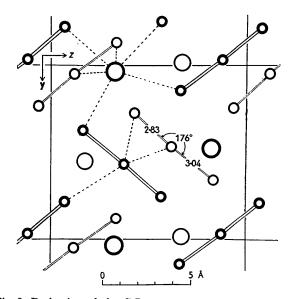


Fig. 2. Projection of the CsI₃ structure on (100). Large and small circles represent cesium and iodine respectively. Heavy lines and circles indicate atoms and bonds in the plane of the drawing, the others in planes 3·43 Å above and below. Intermolecular distances are indicated by dotted lines.

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